TOTAL SYNTHESIS OF (+)-PEDERIN. 2. STEREOCONTROLLED SYNTHESIS OF (+)-BENZOYLSELENOPEDERIC ACID AND TOTAL SYNTHESIS OF (+)-PEDERIN

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Summary: (+)-Benzoylselenopederic acid (1), a left half of (+)-pederin (3), was synthesized stereoselectively based on the $Zn(BH_4)_2$ reduction and total synthesis of (+)-pederin (3) was accomplished from 1 and the previously synthesized 2.

Having completed the stereoselective synthesis of (+)-benzoylpedamide (2), we turned our attention to the synthesis of (+)-benzoylselenopederic acid (1) corresponding to a left half of (+)-pederin (3) and the total synthesis of (+)-pederin therefrom. Retrosynthetic analysis of 1 suggests that the previously developed stereoselective reduction of acyclic ketones with $Zn(BH_4)_2^{-1}$ would be applied successfully in the introduction of chiral C-2 and C-7 hydroxyl groups in (+)-1 as shown in Scheme 1. In fact, $Zn(BH_4)_2$ reduction was proved to be quite effective in the formation of these chiral centers. We now report the highly stereocontrolled synthesis of (+)-pederin (3) from (+)-1 and (+)-2.

Recently, it was reported that optically active α -methyl- β -keto imides or amides were synthesized.² These remarkable findings, coupled with the <u>syn</u>-directing Zn(BH₄)₂ reduction,



made it possible to synthesize optically active $\underline{syn}-\alpha$ -methyl- β -hydroxy acid derivative,² which would be useful as an important building block in the present synthesis as well as in macrolide synthesis. Thus, optically active $(+)-\beta$ -keto imide 4^3 was prepared by Evans' procedure^{2a} and used as a starting material for pederin synthesis. 4 was subjected to reduction with $Zn(BH_4)_2$ in CH_2Cl_2 at -25° to give the desired $\underline{syn}-5^3$ quantitatively with >30 : 1 stereoselection. Imide 5, after protection of the hydroxyl group as the THP ether, was reduced with DIBAH and the resulted aldehyde was treated with lithium enolate of \underline{t} -butyl acetate in THF producing β -hydroxy ester 6 (80%). Reaction of 6 with <u>p</u>-TsOH in MeOH and successive treatment with <u>p</u>-TsOH in refluxing benzene gave α , β -unsaturated lactone 7 (81%).



a) $Zn(BH_4)_2/CH_2Cl_2/-25^\circ$, b) DHP/p-TsOH; DIBAH/PhMe/-78 $^\circ$; LDA/t-BuOAc/THF/-78 $^\circ$, c) p-TsOH/ MeOH; p-TsOH/PhH/reflux,d) MeNO₂/Triton B, e) TiCl₃/Et₃N, f) c. HCl/CH₂Cl₂, g) HSCH₂CH₂SH/ BF₃·Et₂O/CH₂Cl₂/rt, h) LDA/MeOC(Me₂)OCH₂COOMe/THF/-78 -20 $^\circ$; CSA/CH(OMe)₃/CH₂Cl₂/MeOH; DMSO/DCC/Py/CF₃COOH/Et₂O, i) Zn(BH₄)₂/Et₂O/-78 $^\circ$; PhCOCl/DMAP/Py, j) HgO/HgCl₂/aq MeCN; Zn(BH₄)₂/Et₂O, k) PhSeCN/Bu₃P/THF/O $^\circ$, 1) PrSLi/HMPA/rt.

Michael addition of nitromethane to 7 under the conditions reported by Meinwald et al. 4 gave lactone 8^3 having 4 α (equatorial)-CH₂NO₂ group (92%).⁵ The nitromethyl group of 8 was converted into aldehyde with TiCl₃ and Et_3N^6 producing bicyclic lactone 9^3 , which was isomerized exclusively to the more stable lactone 10^3 with c. HCl treatment in CH₂Cl₂ (70% from 8). On treatment with ethanedithiol and BF_3 Et₂0, 10 was converted into thioacetal lactone 11³ (mp 152-152.5°; 81%) having a favourable stereostructure for the next reactions. The isomer 12^3 was obtained from 9 in the same way. 11 was converted into α -keto ester 13 in 3 steps: 1) introduction of glycolic acid moiety (LDA/MeOC(Me₂)OCH₂COOMe); 2) conversion of the resulted hemiacetal into the methyl ether and simultaneous deprotection of methoxy i-propyl ether (CSA/CH(OMe)₃/MeOH); 3) Moffatt oxidation (DMSO/DCC); 91% overall yield. $Zn(BH_A)_2$ reduction of 13 in Et₂O at -78° yielded the desired 7 β -alcohol 14 quantitatively with 17 : 1 stereoselection. Benzoylation of 14 afforded 15 (mp 135-6°; 86%), whose stereostructure was confirmed by X-ray crystallography. Dethioacetalization of 15 followed by $Zn(BH_4)_2$ reduction produced alcohol 16 in 91% yield. Phenylselenation of 16 was accomplished by PhSeCN and Bu_3P treatment⁷ giving 17^3 (94%) without hydrolyzing benzoate group. Treatment of 17 with NaOMe gave alcohol 18, which was converted into (+)-methyl pederate (19) by deselenation. ¹H NMR data of 18 and 19 were identical with those of the corresponding authentic samples.⁸ Selective hydrolysis of methyl ester of 17 was accomplished by PrSLi treatment in HMPA⁹ giving (+)-1 quantitatively.

Then, we examined the total synthesis of (+)-pederin (3) from (+)-1 and (+)-2. Coupling of both segments was carried out in a similar manner reported by Matsumoto et al.¹⁰; reaction of acid chloride 20 obtained from 1 and imidate 21 from 2 gave N-acylimidate 22 which was immediately reduced with NaBH₄. Reduction in EtOH afforded 23^3 (19%) and 10α -isomer 24 (45%).



a) SOC1₂/Py/CH₂C1₂, b) Me₃OBF₄/CH₂C1₂, c) Et₃N/CH₂C1₂, d) NaBH₄, e) <u>i</u>-PrSH/CSA/CH₂C1₂, f) HgC1₂/Et₃N/MeOH, g) NaIO₄; Et₃N/PhH; 1N LiOH/MeOH.

whereas reduction in <u>i</u>-PrOH and CH_2Cl_2 produced 23 (28%) and 24 (30%), the selectivity being slightly improved. Epimer 24 was successfully converted into 23 as follows.¹¹ Treatment of 24 with <u>i</u>-PrSH and CSA in CH_2Cl_2 gave thioacetal 25 which was treated with $HgCl_2$ in MeOH in the presence of Et_3N^{12} to give 23 in 47% yield (from 24) along with epimer 24 (36%). ¹H NMR data of 23 were identical with those of the authentic sample.¹⁰ As 23 has already been converted into (+)-pederin (3),¹⁰ total synthesis of 3 was thus accomplished.

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References and Notes

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- 3. ¹H NMR spectra were taken on a JEOL GX-400 instrument in CDCl₃. 4: NMR & 2.33 (s; MeCO), 4.51 (q, \underline{J} =7.3 Hz; 3-H); $[\alpha]_0^{20}$ +26.5°, $[\alpha]_0^{35}$ +22.9° (c=1.4, CH₂Cl₂). 5: NMR & 1.17 (d; \underline{J} =7.1 Hz; Me), 1.21 (d, \underline{J} =6.4 Hz; Me); $[\alpha]_0^{26}$ -71.8° (c=1.61, CHCl₃). 8: NMR & 2.42 (dd, \underline{J} =16.0, 10.0 Hz; 5 α -H), 2.71 (dd, \underline{J} =16.0, 6.4 Hz; 5 β -H); $[\alpha]_0^{25}$ +74.3° (c=1.11, CHCl₃). 9: NMR & 2.63 (dd, \underline{J} =19.0, 8.7 Hz; 5-Ha), 2.65 (dd, \underline{J} =19.0, 4.9 Hz; 5-Hb). 10: NMR & 2.45 (dd, \underline{J} =18.1, 3.9 Hz; 5-Ha), 2.88 (dd, \underline{J} =18.1, 10.7 Hz; 5-Hb). 11: NMR & 2.26 (dd, \underline{J} =18.1, 12.0 Hz; 5 β -H), 2.90 (ddd, \underline{J} =18.1, 5.6, 1.2 Hz; 5 α -H), 4.33 (d, \underline{J} =10.0 Hz; SCHS); $[\alpha]_0^{26}$ -12.4° (c=1.45, CHCl₃). 12: NMR & 2.57 (dd, \underline{J} =15.9, 10.8 Hz; 5 α -H), 2.73 (dd, \underline{J} =15.9, 6.4 Hz; 5 β -H), 4.76 (d, \underline{J} =5.4 Hz; SCHS). 17: NMR & 3.24, 3.78 (each s; 2xOMe), 5.37 (s; 7-H); $[\alpha]_0^{26}$ +91.9° (c=3.6, CHCl₃). 23: NMR & 5.31 (dd, \underline{J} =9.8, 3.7 Hz; 10-H), 5.44 (s; 7-H), 6.69 (d, \underline{J} =9.8 Hz; NH); $[\alpha]_0^{26}$ +67.1° (c=1.55, CHCl₃).
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- F. Matsuda, M. Tomiyoshi, M. Yanagiya, and T. Matsumoto, <u>Tetrahedron Lett.</u>, <u>24</u>, 1277 (1983).
- 11. Matsumoto et al. extensively studied the epimerization of epi-24 to 23 by acid catalyzed double alkoxy-exchange reaction (see ref. 10).
- cf. F. Nakatsubo, A. J. Cocuzza, D. E. Keeley, and Y. Kishi, <u>J. Am. Chem. Soc</u>., <u>99</u>, 4835 (1977).

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